Self-organization and phoretic motions for the origin of life

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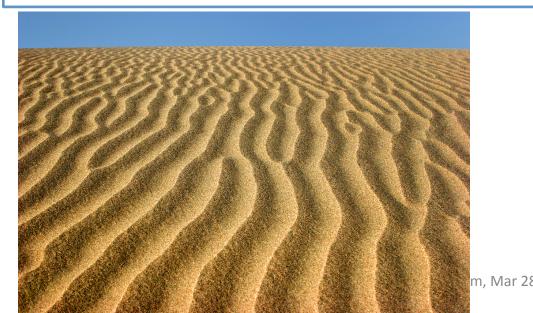
Self-organization

Equilibrium system

- Second law of thermodynamics tells entropy in a closed and isolated system never decreases.
- The system evolves to equilibrium of maximum entropy.

Non-equilibrium system

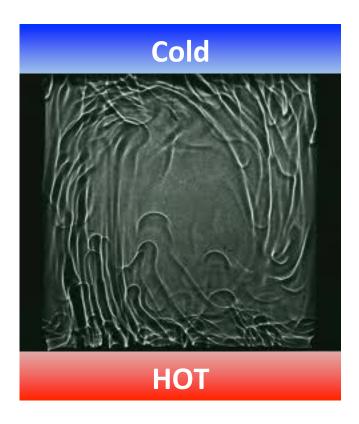
- The system develops towards a self-organized state.
- The flow of mass or energy keeps system organized.





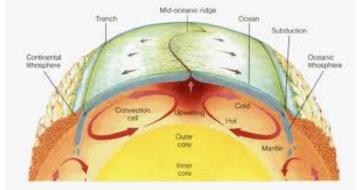
Non-equilibrium physics and the earth

Thermal convection

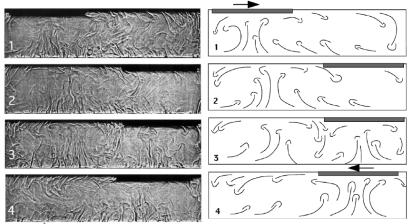


Self-organization (non-equilibrium process)
 is essential for the earth's environment.

Plate tectonics and mantle convection in the earth



Tectonics demonstration in a lab scale



J Zhang and A Libchaber Phys Rev Lett 84, 4361 (2000)

Self-organization behind origin of cellular life



The origin of life

A B

Micelle

Liposome

Bilayer sheet

C

2

3

4

To obtain self-organized biological systems

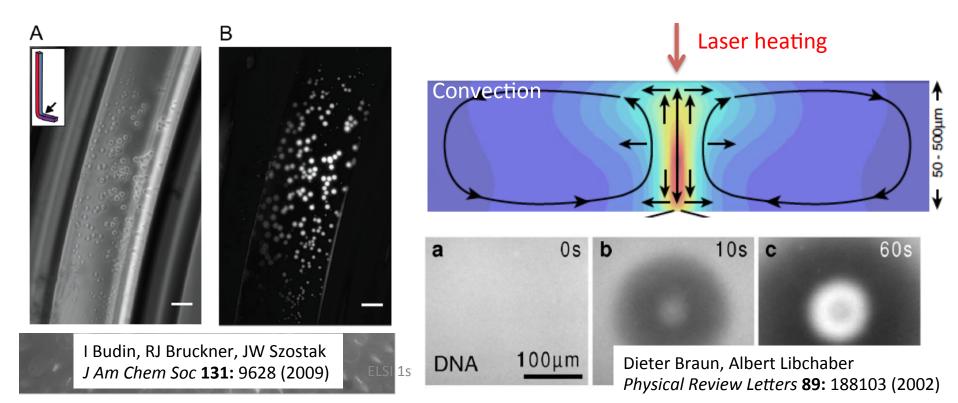
- ⇒ Molecules should be concentrated above critical concentration.
- → Molecules should be selected for proper function.

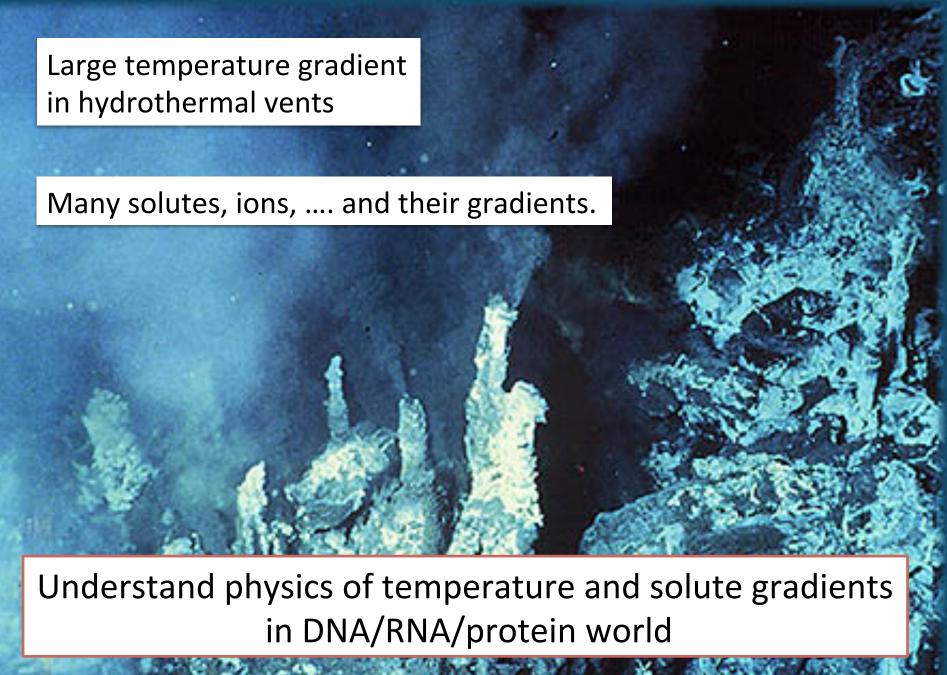
Did non-equilibrium processes drive (accelerate) the origin of life?

Thermal convection and the origin of life

Thermal convection: well-known non-equilibrium system

- Temperature gradients drive the flow of water.
- Thermal convection occurs if Reynolds number is large.
- Thermal convection traps DNA and lipid vesicle. → accumulation

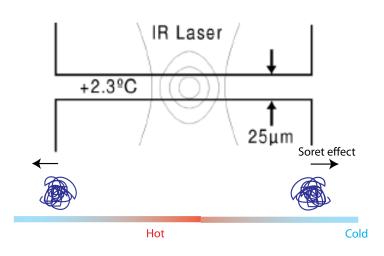




The Soret effect: Transport under a temperature gradient

The Ludwig-Soret effect: molecular transport discovered more than 100 years ago

- Molecules move with a velocity proportional to temperature gradient.
- Typical charged molecules, DNA, RNA, Proteins, move to Cold from Hot.

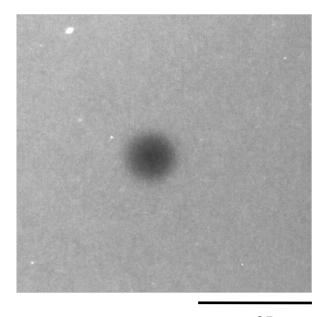


Velocity of thermal diffusion

$$v = -D_T \nabla T$$

Density flow of DNA

$$J = -D\nabla c - cD_{T}\nabla T$$

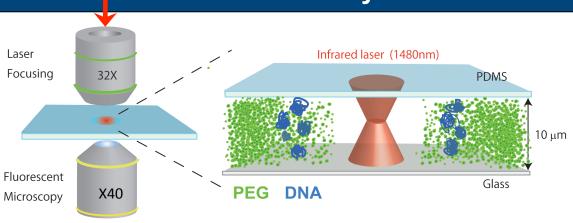


35 µm

D Braun and A Libchaber *Phys Rev Lett* **89:** 188103 (2002)

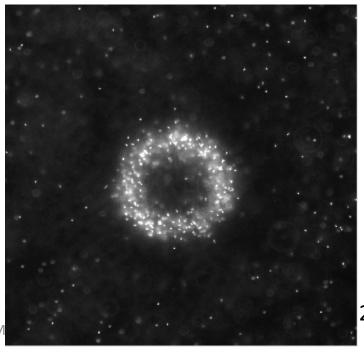
ELSI 1st symposium, Mar 28, 2013

The Soret effect can trap and manipulate microsize objects



Accumulation of fluorescent beads (0.5 μm)

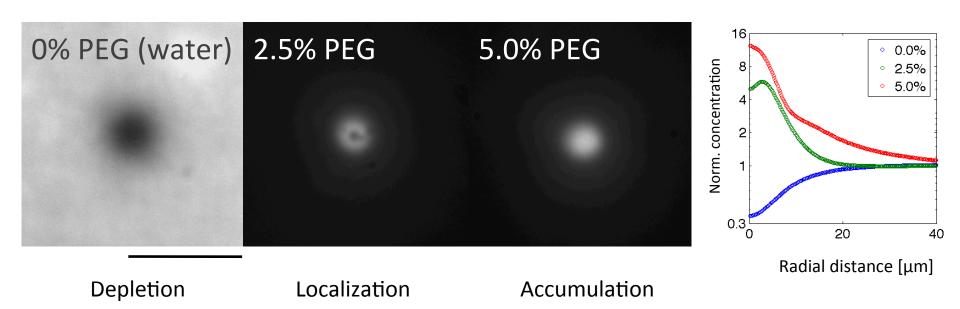
Ring of fluorescent beads



5%PEG

posium, N

DNA 5.6kbp (ϕ =0.01%)



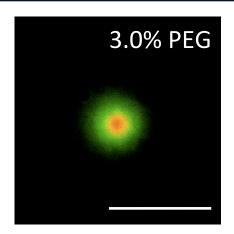
H.R.Jiang, H.Wada, N.Yoshinaga, M.Sano *Physical Review Letters* **102**: 208301 (2009)

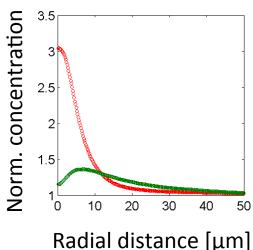
Yusuke T. Maeda, Axel Buguin, Albert Libchaber *Physical Review Letters* **107**: 038301 (2011)

Size-dependent localization and separation

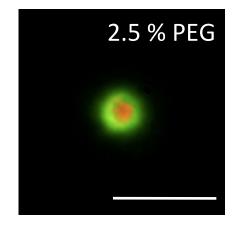
The separation of

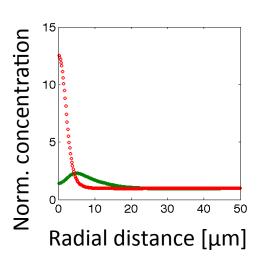
Long RNA (1.5 kb, ϕ =0.01%)/ short DNA (0.25 kbp, ϕ =0.01%)





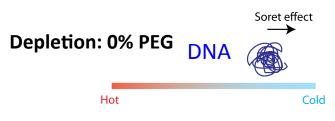
Large colloid (0.5 μ m, ϕ =0.01%) / Small colloid (0.1 μ m, ϕ =0.01%)





Diffusiophoresis model agrees with experimental observation

PEG gradient generates the osmotic force (diffusiophoresis) on DNA surface.

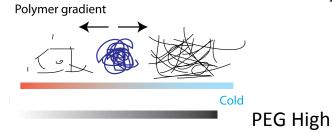




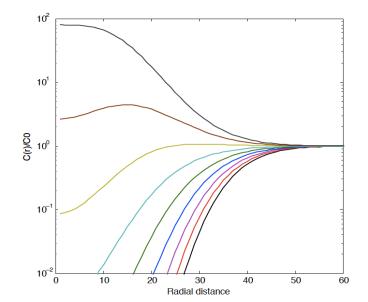
Normal Diffusion

The Soret effect

Diffusiophoresis

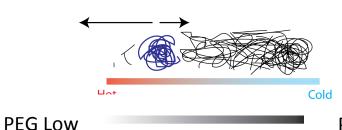


 $c_{DNA}(r) = c_{DNA}^{0} \exp \left[-S_{T}(T(r) - T_{0}) + \left(c_{PEG}^{0} - c_{PEG}(r) \right) V \right]$



Accumulation: >3% PEG

PEG Low



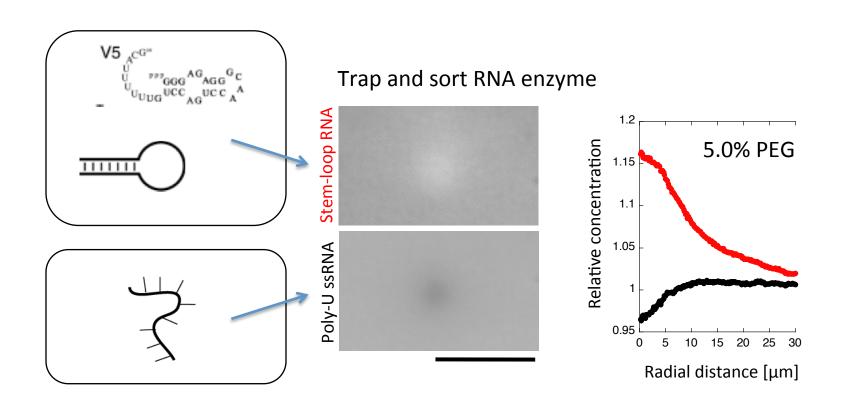
PEG High

J.L. Anderson, Ann. Rev. Fluid. Mech **21**, 61 (1989)

ELSI 1st symposium, Mar 28, 2013

Jiang, et al PRL (2009); Maeda, et al PRL (2011)

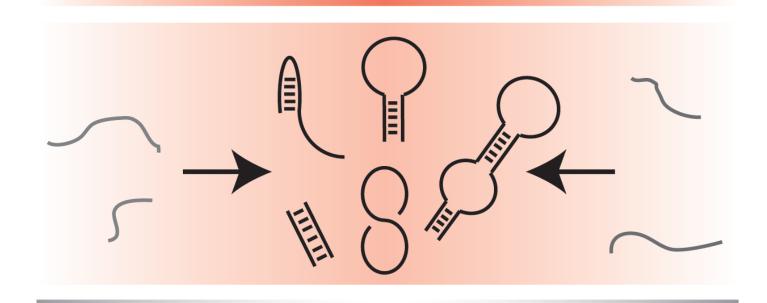
Folding dependent trapping of small RNA



- Stem DNA/RNA, RNA enzyme that has stem longer than 6-bp: Accumulation
- ssDNA/ssRNA up to 120 nt: Depletion

YT Maeda, T Tlusty, A Libchaber Proc Natl Acad Sci USA 109: 17972 (2012) CJ Wienken, P Baaske, S Duhr, D Braun Nucleic Acid Research 39, e52 (2011)

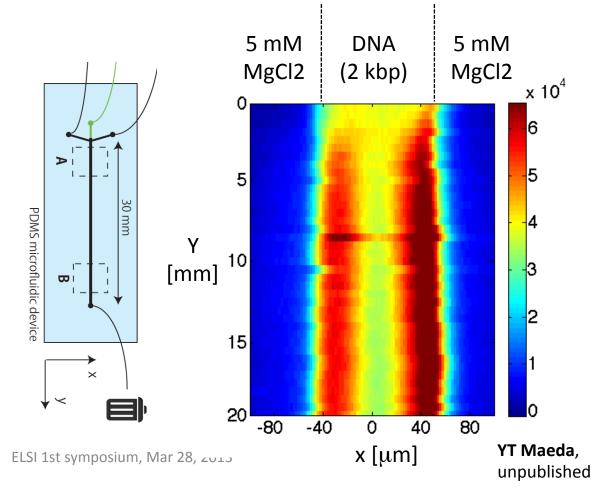
Temperature Gradient



Entropic Force Gradient

- One may criticize PEG is artificial.
- Most abundant solute in nature is ions.
- Ions, e.g. Mg, can do as well as PEG.

THE RESERVE OF THE PERSON NAMED IN	Fluid	Seawater
Temperature (C)	360-365	2
Acidity (at 25 C)	3.35	7.8
Dissolved Oxygen	0	0.076
Hydrogen Sulfide (mM)	23-35	.0
Sodium (mill)	537	464
Potassium (mM)	17.1	9.8
Calcium (nM)	30.8	10.2
Magnesium (mM)	0	52.7
Silica (mM)	20.75	0.2
Chloride (mM)	636	541
Sulfate (mM)		27.9
Manganese (µM)	680	0
Iron (µM)	5590	0.0015
Copper (µM)	98-120	0.007
Zinc (µM)	47-53	0.01



The Soret effect and isotope fractionation

JUNE 1, 1939

PHYSICAL REVIEW

VOLUME 55

On the Theory of Isotope Separation by Thermal Diffusion

W. H. Furry, R. Clark Jones, Research Laboratory of Physics, Harvard University, Cambridge, Massachusetts

And

L. Onsager, Department of Chemistry, Yale University, New Haven, Connecticut (Received April 6, 1939)

- The Soret effect was studied for isotope separation.
- Recent studies show isotope fractionation through the Soret effect in silica melts.

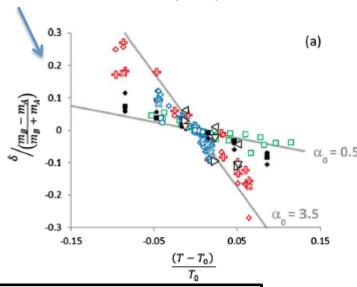
DJ Lacks, et al.

Phys Rev Lett 108: 065901 (2012)

F Huang, et al.

Nature 464: 396 (2010)

Non-zero means isotope separation



Non-equilibrium transport could be related to both the earth and life.

Summary and future problems

- DNA/RNA as a solute moves along a temperature gradient. It is called Soret effect.
- Non equilibrium process, the Soret effect, leads accumulation and separation of DNA/RNA.
- Accumulation and selection of RNA enzymes may accelerate the birth of RNA world.

Future challenges in the origin of life out of equilibrium

- Evolution and natural selection through non-equilibrium process
- Replications of DNA and RNA far from equilibrium
- Origin of genetic code

DNA replication under convection D Braun, N Goddard, A Libchaber Phys Rev Lett **91**: 158103 (2003)

