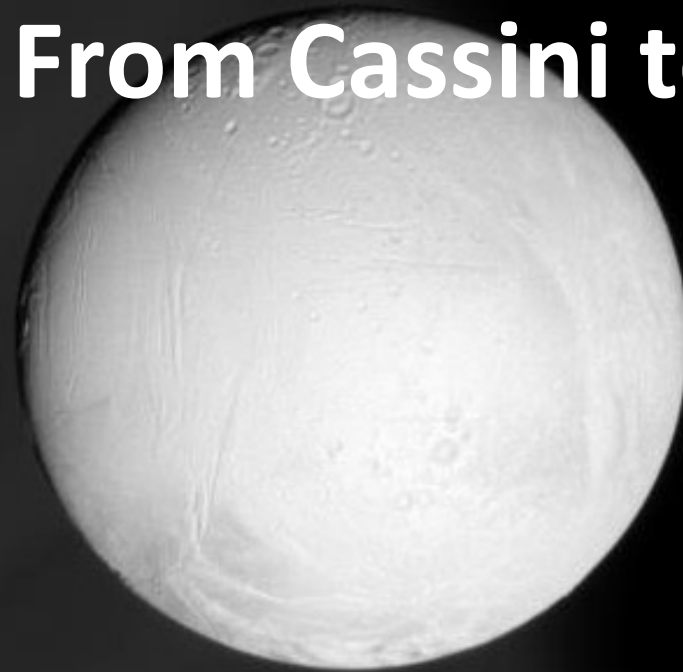


**Cosmochemistry  
geochemistry, & astrobiology  
on icy satellites:  
From Cassini to JUICE**



Yasuhito Sekine  
Univ. Tokyo  
ELSI kick-off meeting,  
29 March 2013

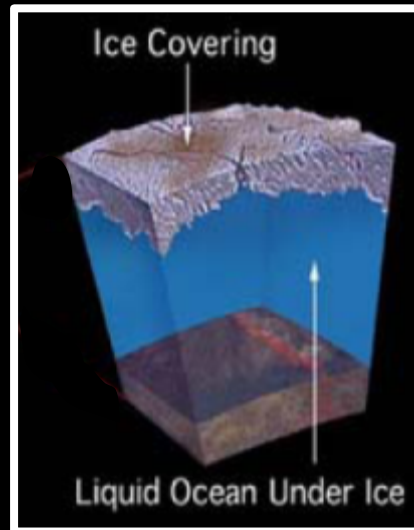
# What are icy satellites?

- **Ice-covered moons** around giant planets
- **Wide variety** in activity, surfaces, & interiors

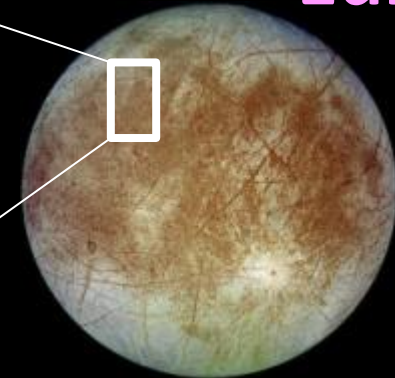
**Callisto**



**Low geological activity**

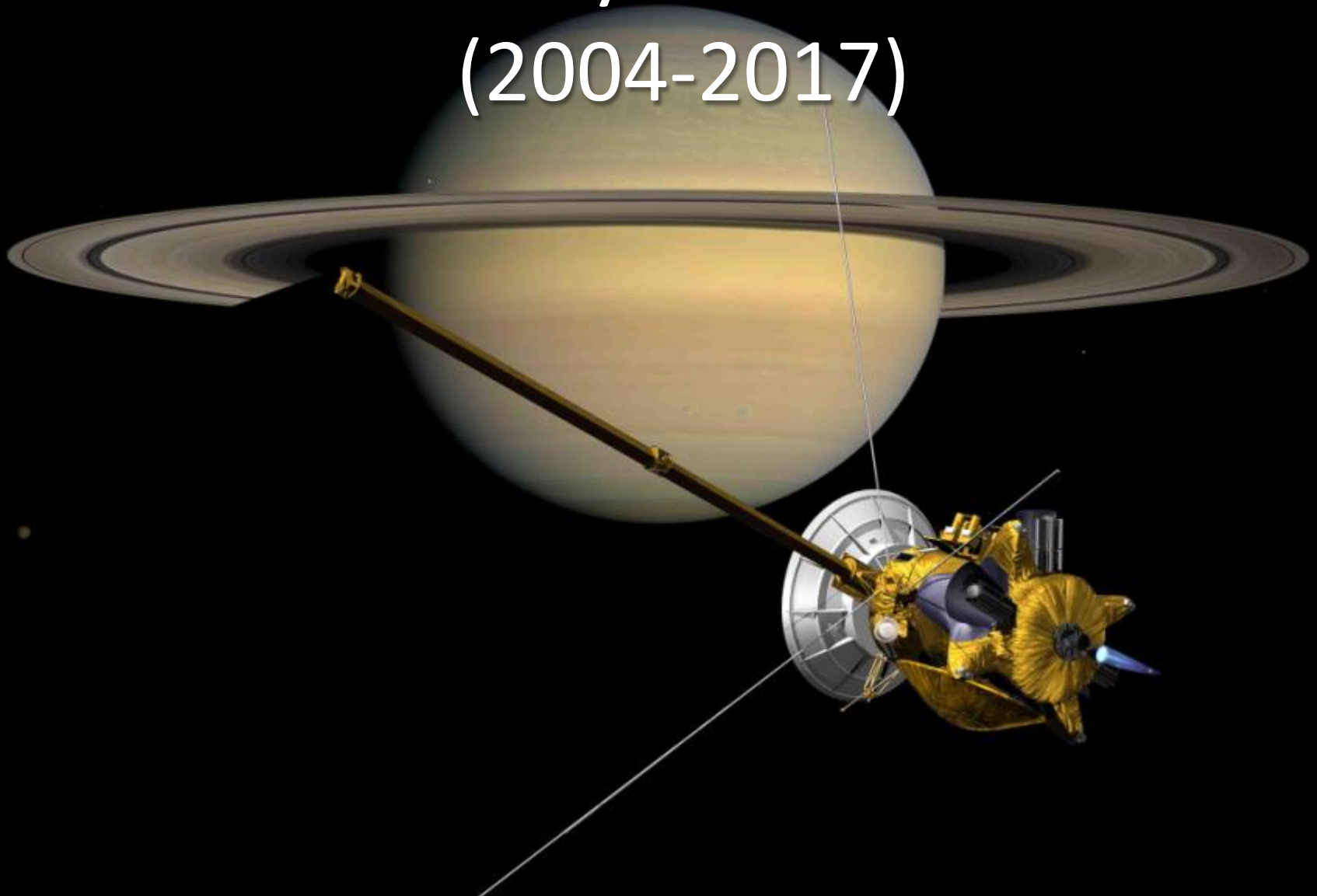


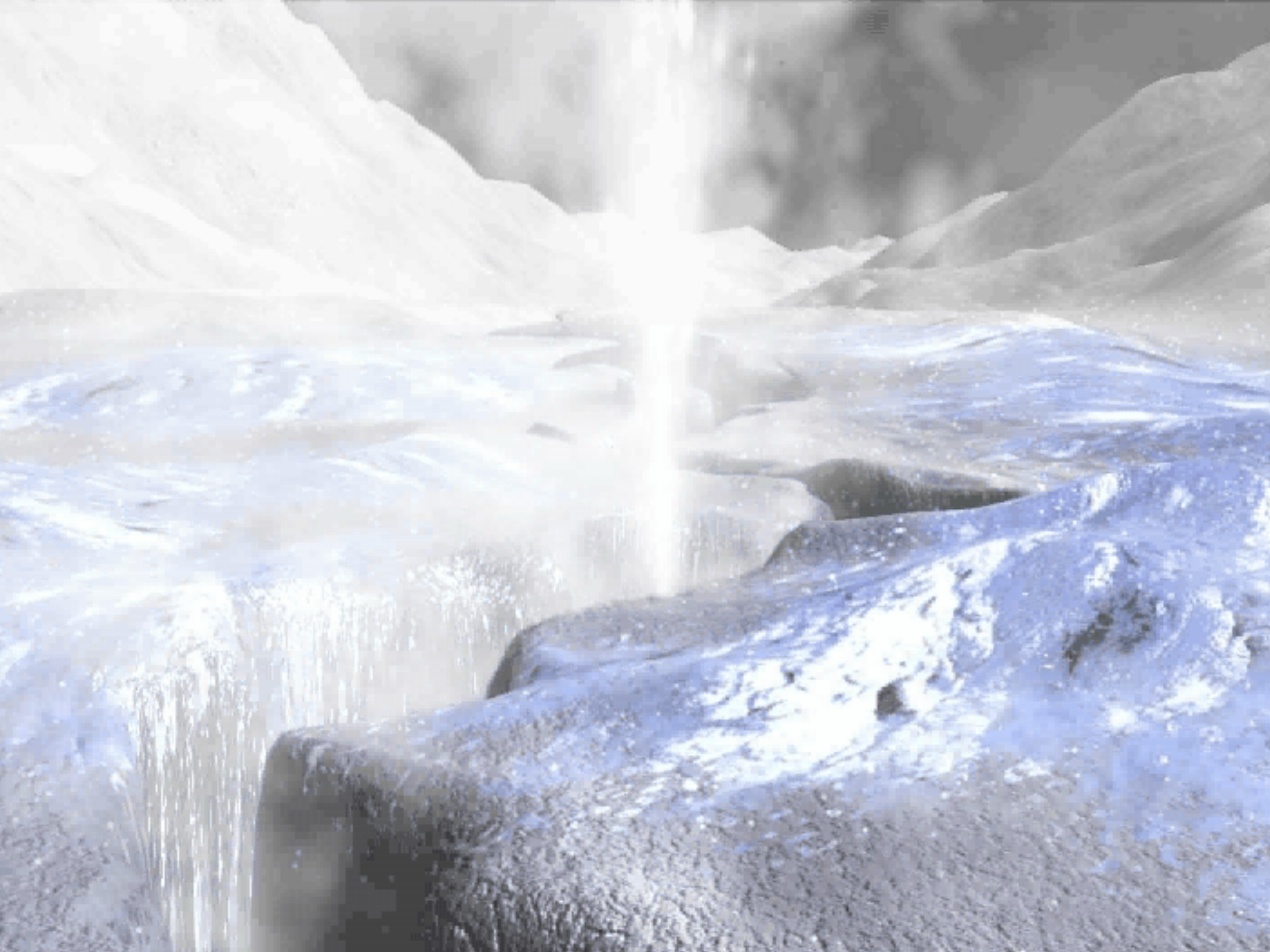
**Europa**



**Extensive & prolonged activity, interior ocean**

# Cassini mission to Saturnian system (2004-2017)

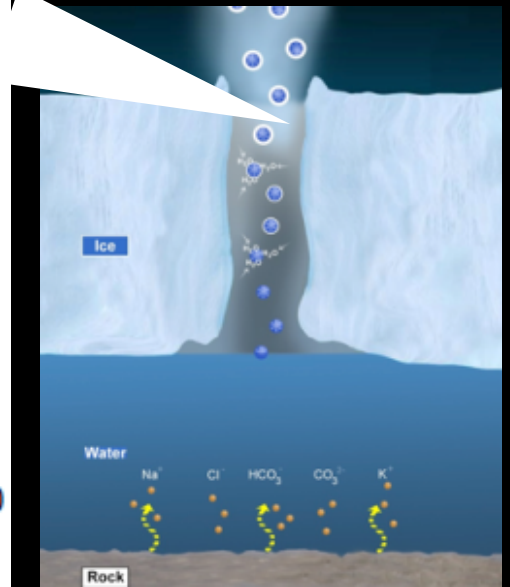
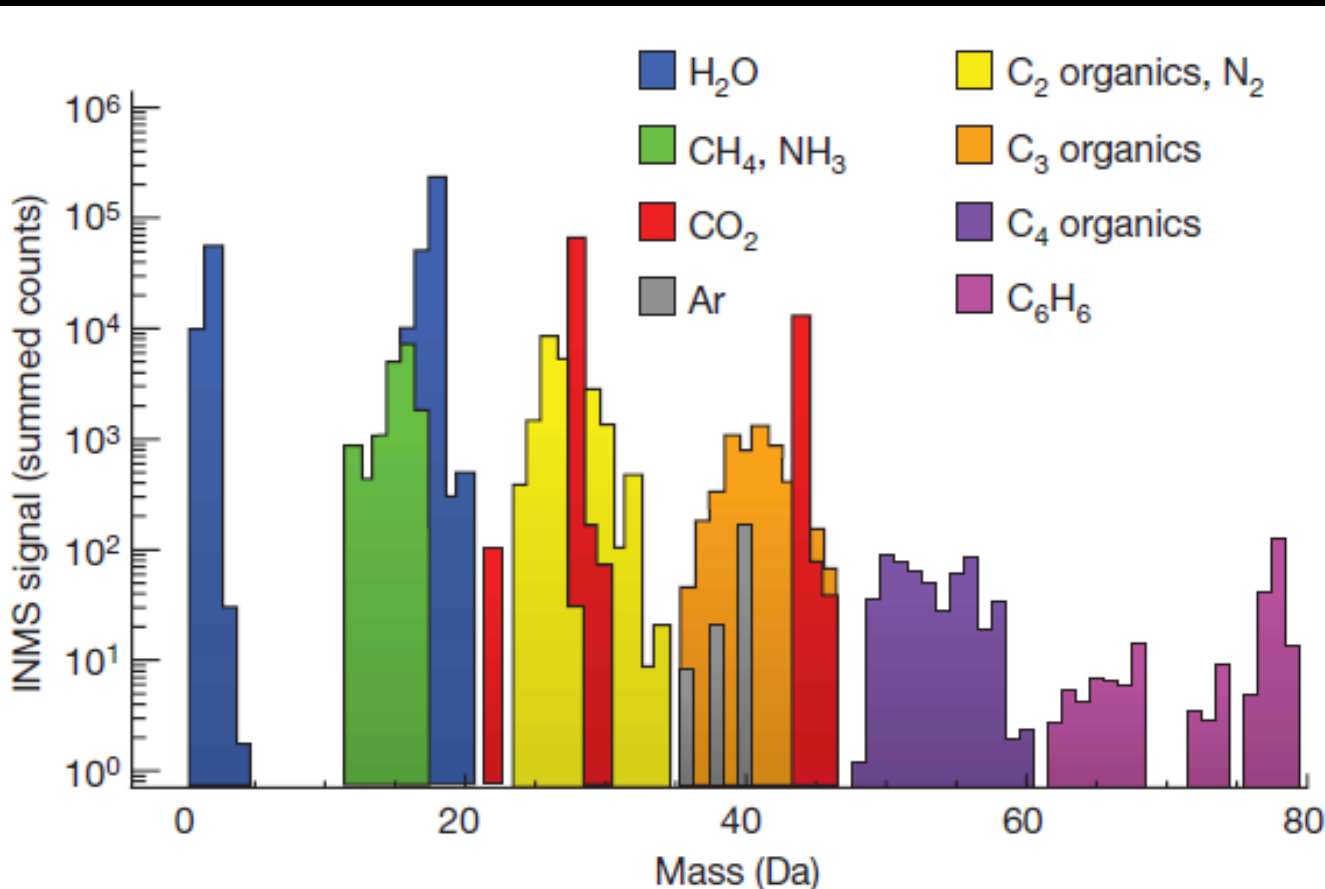




# Mass spectra of Enceladus' plume

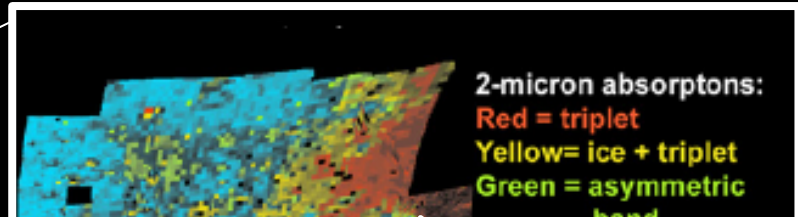
**Gas:** H<sub>2</sub>O + C-species (CO<sub>2</sub> (1-5%), CH<sub>4</sub> (1%), alcohols, aldehydes) + N-species (NH<sub>3</sub> (1%), HCN) (Waite et al. 2009)

**Solids:** H<sub>2</sub>O ice with Na-salts, carbonates (Postberg, 2009; 2011)

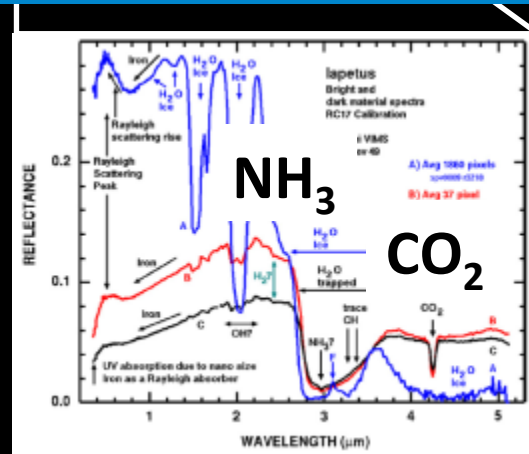
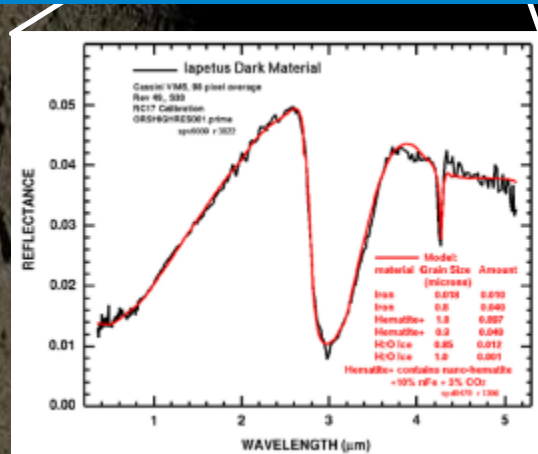
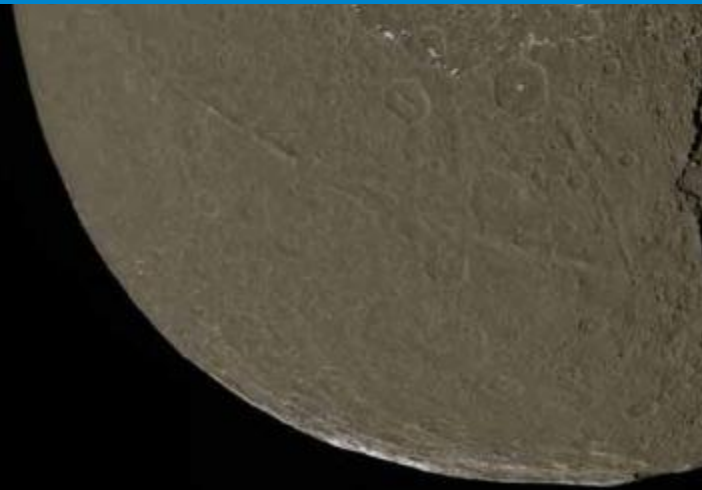




# Mapping of ices & organics by Cassini



Cassini has opened an important window into the chemistry of icy satellites



Clark et al., 2012

Iapetus: a Saturn's mid-sized moon

# Why think about chemistry?

Present-day composition =

**initial conditions** + **subsequent evolution**

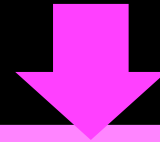


- Protoplanetary disk
- Giant planet formation

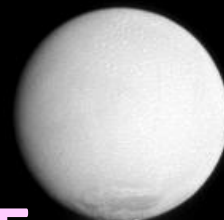


**Callisto: "fossils"**

Satellites that are not active



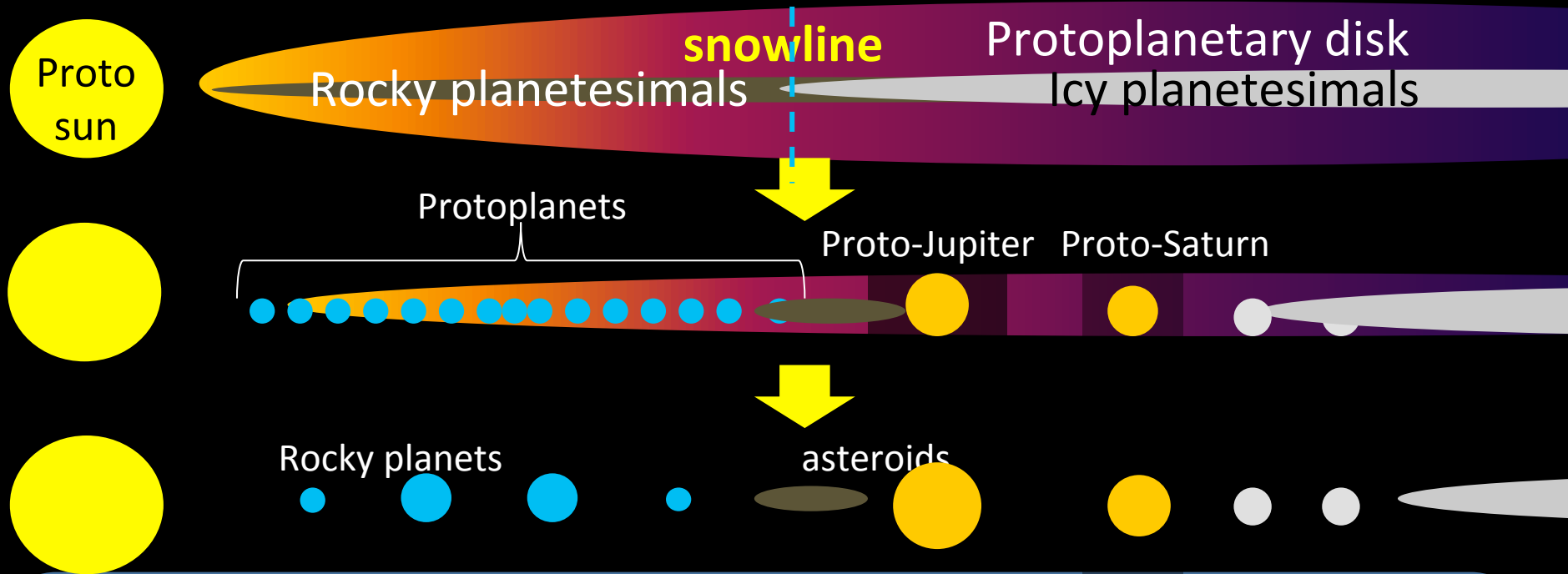
- Geological processes
- Geochemical products



**Europa, Enceladus**

Active satellites

# Formation of solar system & icy satellites



## Giant planet formation (e.g., Canup & Ward, 2002)

1. Formation of proto-planet in a disk
2. Gas & solid accretion & a sub-disk around planet
3. Satellite formation

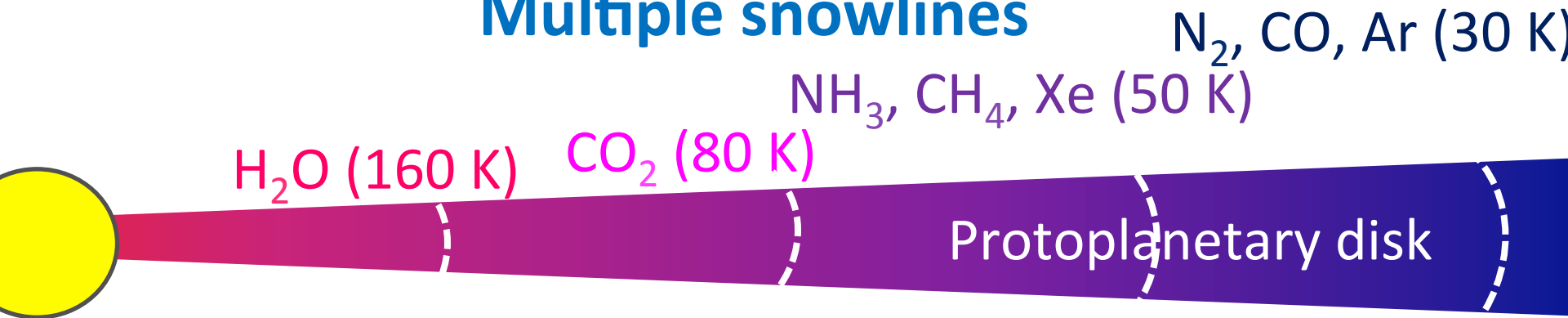
Initial comp.  $\Leftrightarrow$  solids in protoplanetary disk/sub-disk



# Ice composition: proxy for temperature

**Trapping of gas into ice:** Gas species are trapped in solids, but the condensation **temperature** depends on the gas species (e.g., Hersant et al., 2004; Alibert & Mousis, 2007)

## Multiple snowlines

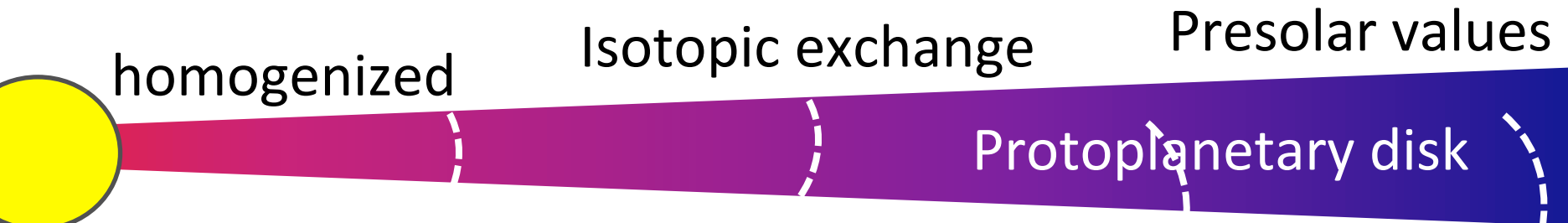


**Cassini: Enceladus & Titan** → lack <sup>36</sup>Ar, lots CH<sub>4</sub> & NH<sub>3</sub>  
→ **disk temp @ Saturn: ~30–50 K** (Mousis et al., 2009)

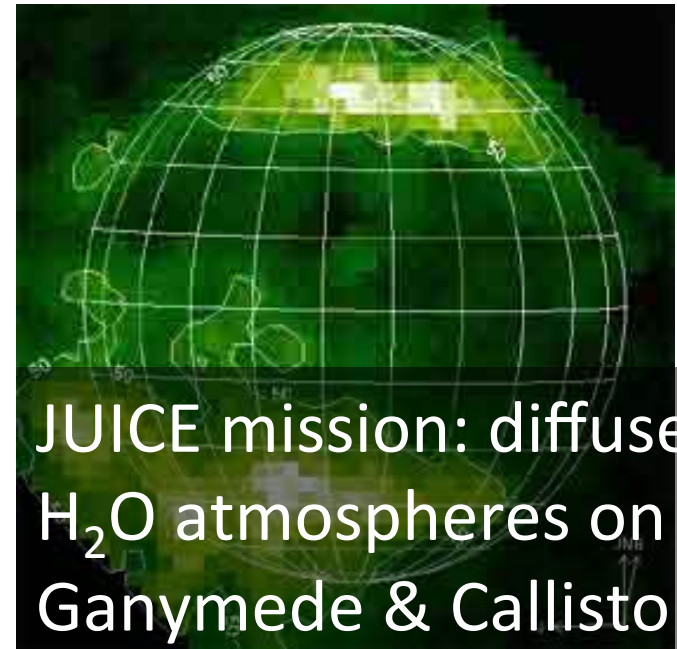
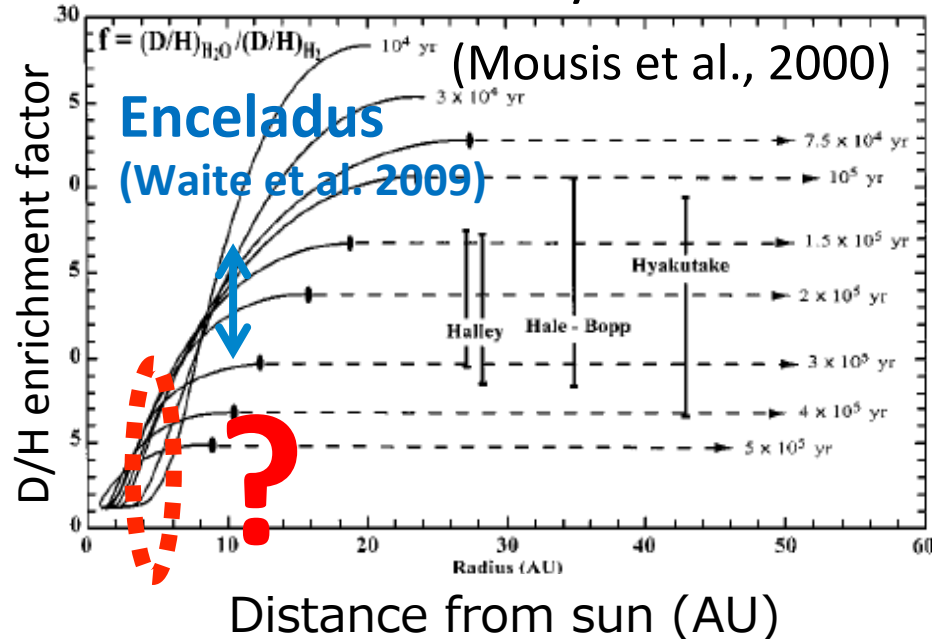
**JUICE mission: High-resolution observations of volatiles on Jupiter's moons**

# Isotopes of volatiles: D/H of water

- **D/H ratio of H<sub>2</sub>O: Isotope exchange** between H<sub>2</sub>O & H<sub>2</sub> occurred as a function of **disk temperature, viscosity, & timing of gas dissipation** (e.g., Drouart et al., 1999; Mousis et al., 2000)



Coefficient of viscosity  $\alpha \sim 0.003$



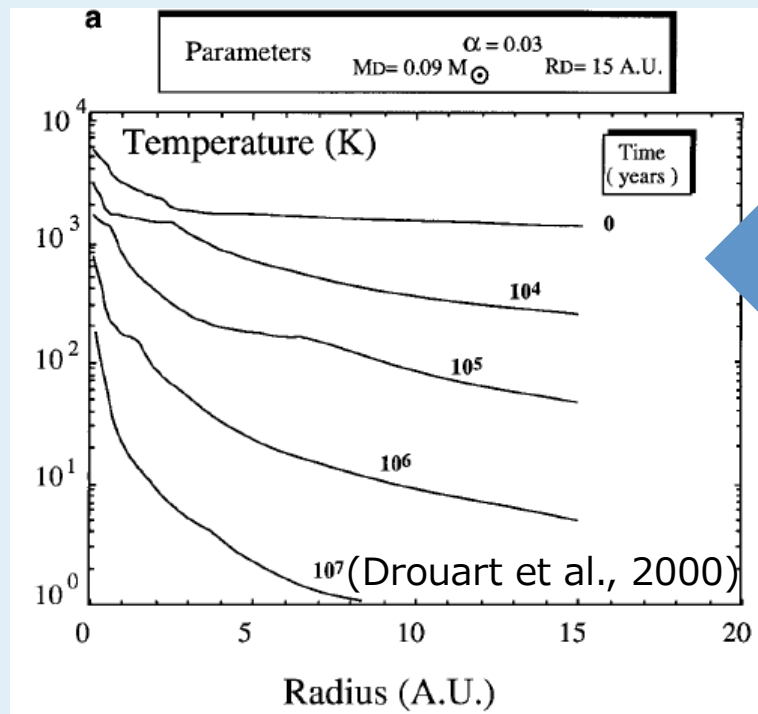
JUICE mission: diffuse H<sub>2</sub>O atmospheres on Ganymede & Callisto

# Characterization of our solar system

## Theoretical disk models

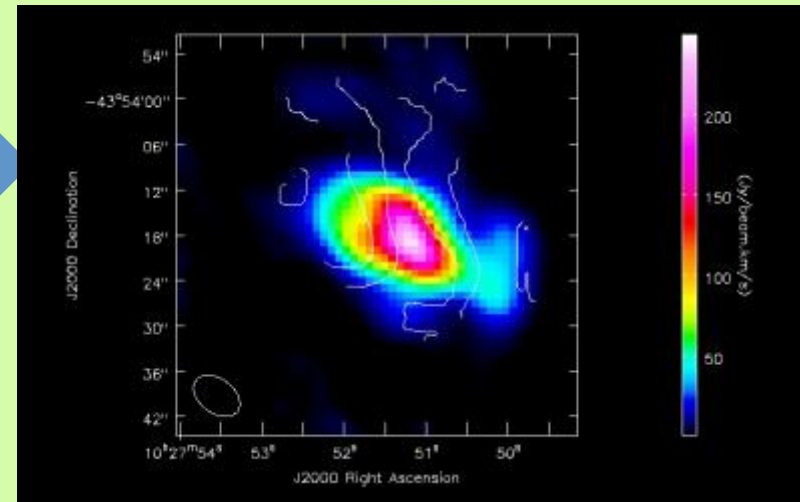
Temperature profile of our disk

Building materials of Earth



## High-resolution observations (ALMA, TMT)

Structure of protoplanetary disks  
How common is the initial condition



High-resolution observation of protoplanetary disks (c) ALMA

Exploration of the solar system: Factual evidence to characterize our solar system

# Why think about chemistry?

Present-day composition =

**initial conditions** + **subsequent evolution**

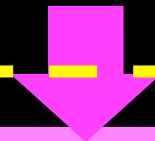


- Protoplanetary disk
- Giant planet formation

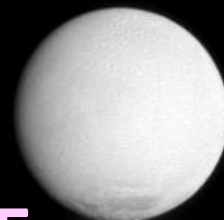


**Callisto: "fossils"**

Satellites that are not active



- Geological processes
- Geochemical products



**Europa, Enceladus**

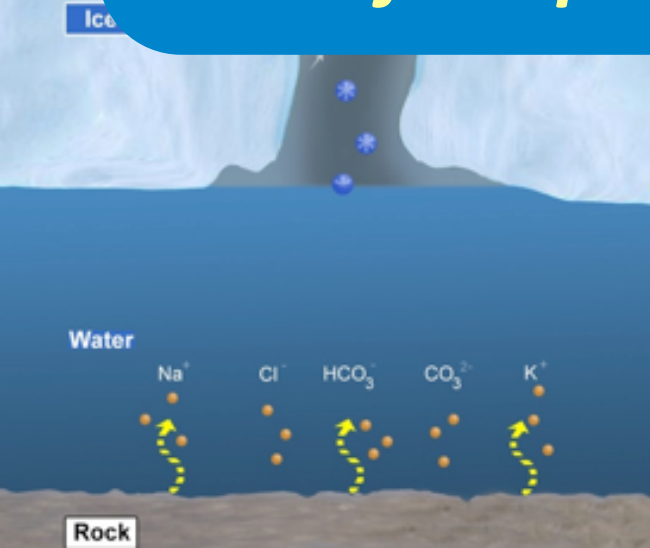
Active satellites

# Enceladus' plume & interior ocean

## *Primary questions*

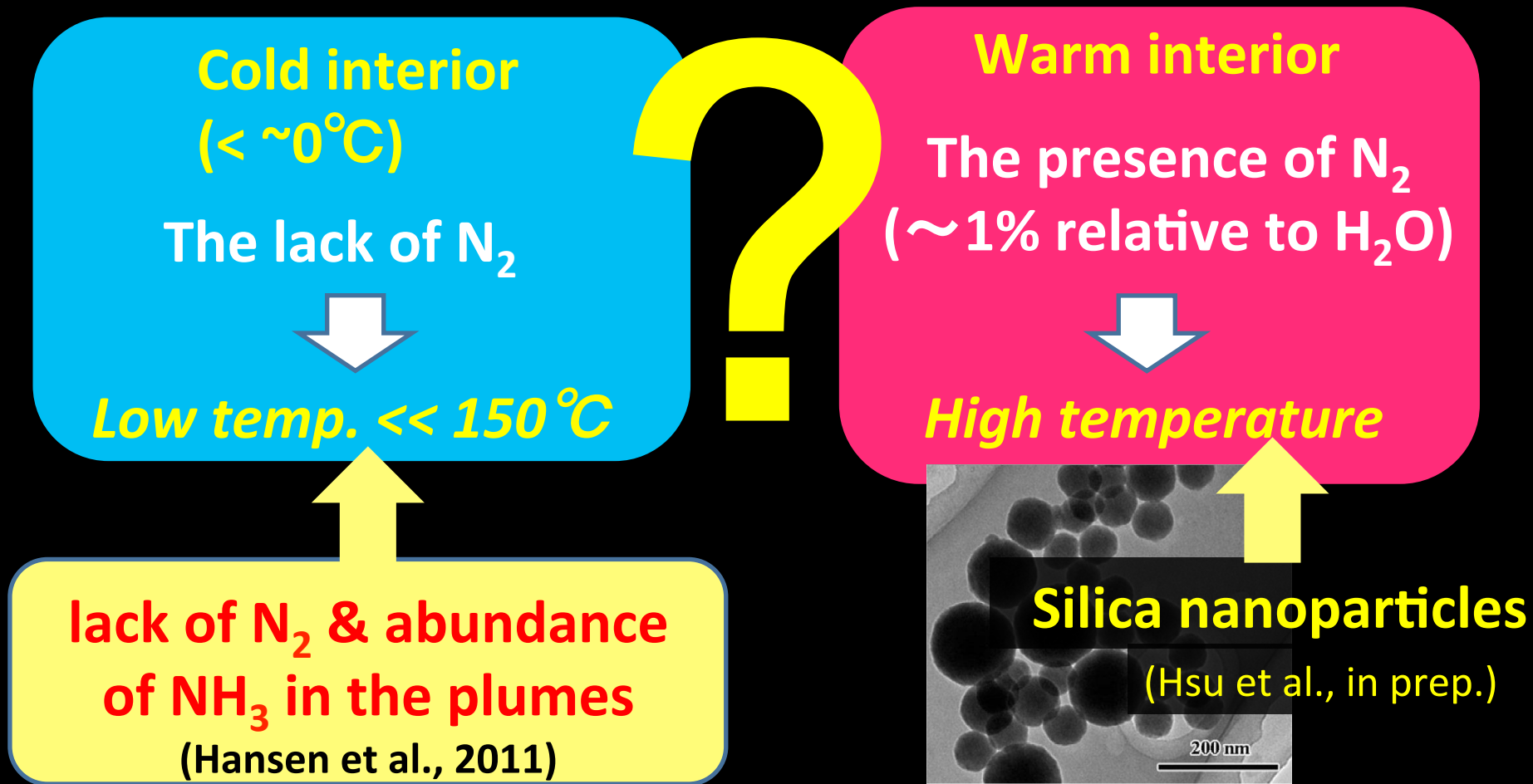
*Are there any hydrothermal systems?*

*What are particular conditions of temperature  
fluid pH, and rock components?*



# Cold or hot in Enceladus?

- If hot,  $\text{N}_2$  should be observed in the plumes formed by thermal dissociation of  $\text{NH}_3$ :  $2\text{NH}_3 \Rightarrow \text{N}_2 + 3\text{H}_2$  (Matson et al. 2007).





# Experimental

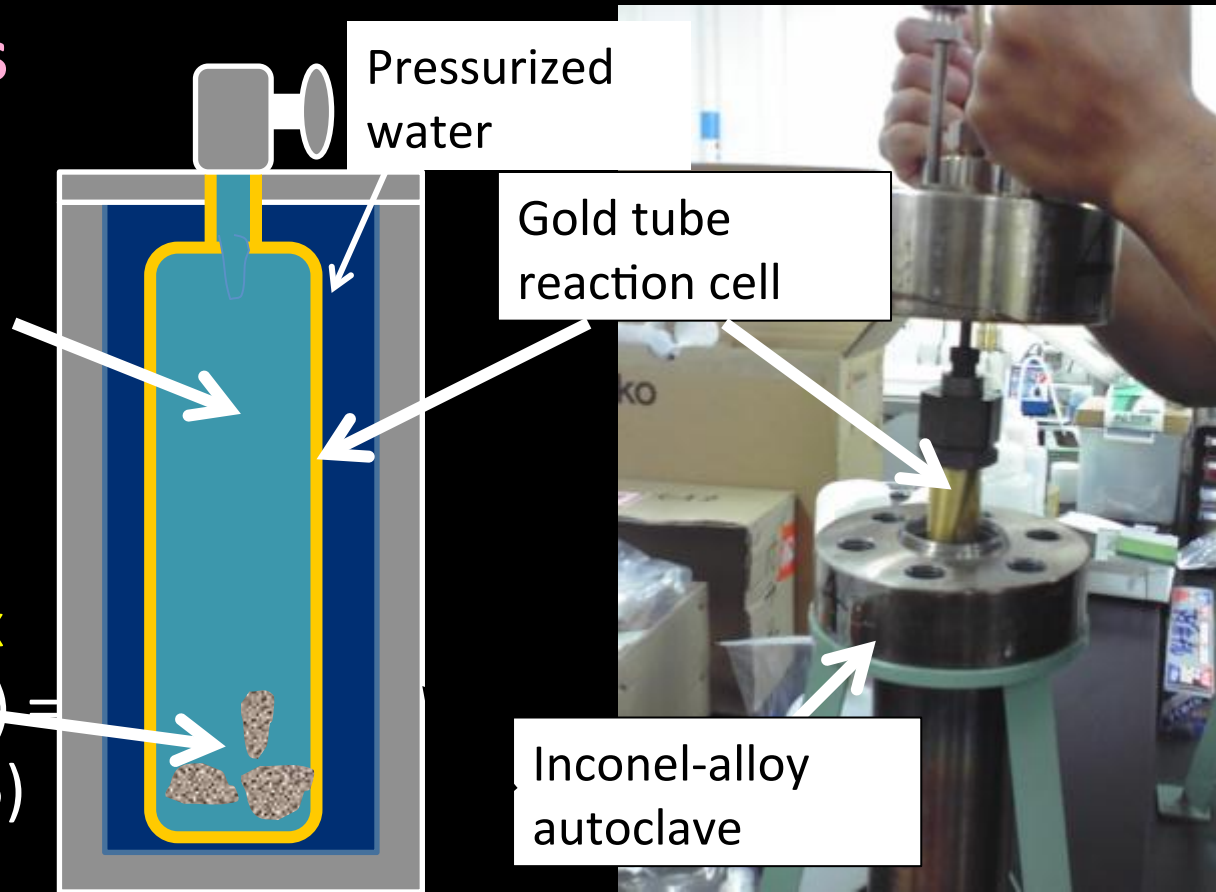
Enceladus interior ~ Earth's sea floors

$P = 100\text{--}400\text{ bar}$  &  $T < 400^\circ\text{C}$

- **Starting materials**

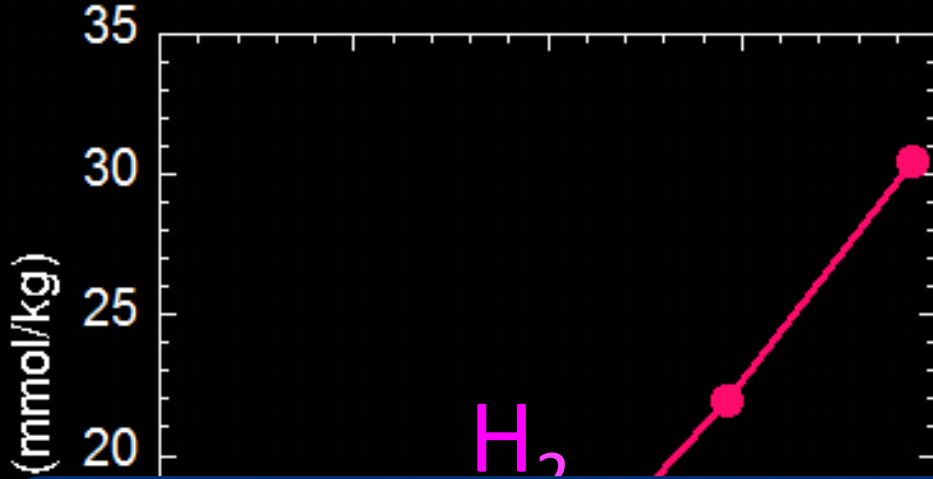
An aqueous solution of  
 $\text{NH}_3$  (1%) &  $\text{NaHCO}_3$   
( $\text{CO}_2$ ) (0.1% or 3%)  
(pH = 10)

Powdered olivine + opx  
(Mg# (=Mg/(Mg+Fe)) =  
90) (rock:water = 1:5)



# Results: at 300°C

- Dissolved gas species

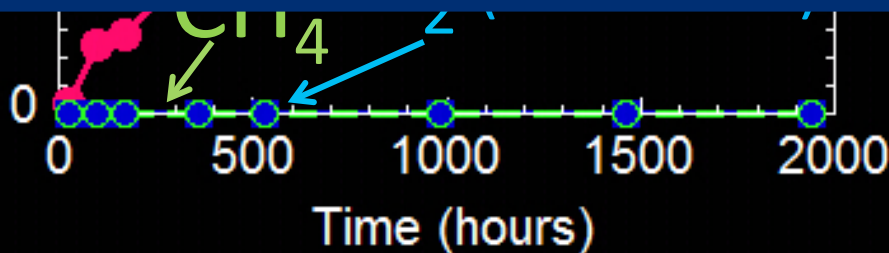


- Mineralogical analyses for the rock material

65% of olivine was converted to serpentine & magnetite



NH<sub>3</sub> dissociation is kinetically inhibited due to high activation energy and high concentrations of H<sub>2</sub>



magnetite

# Results: pH, silica, & temperature

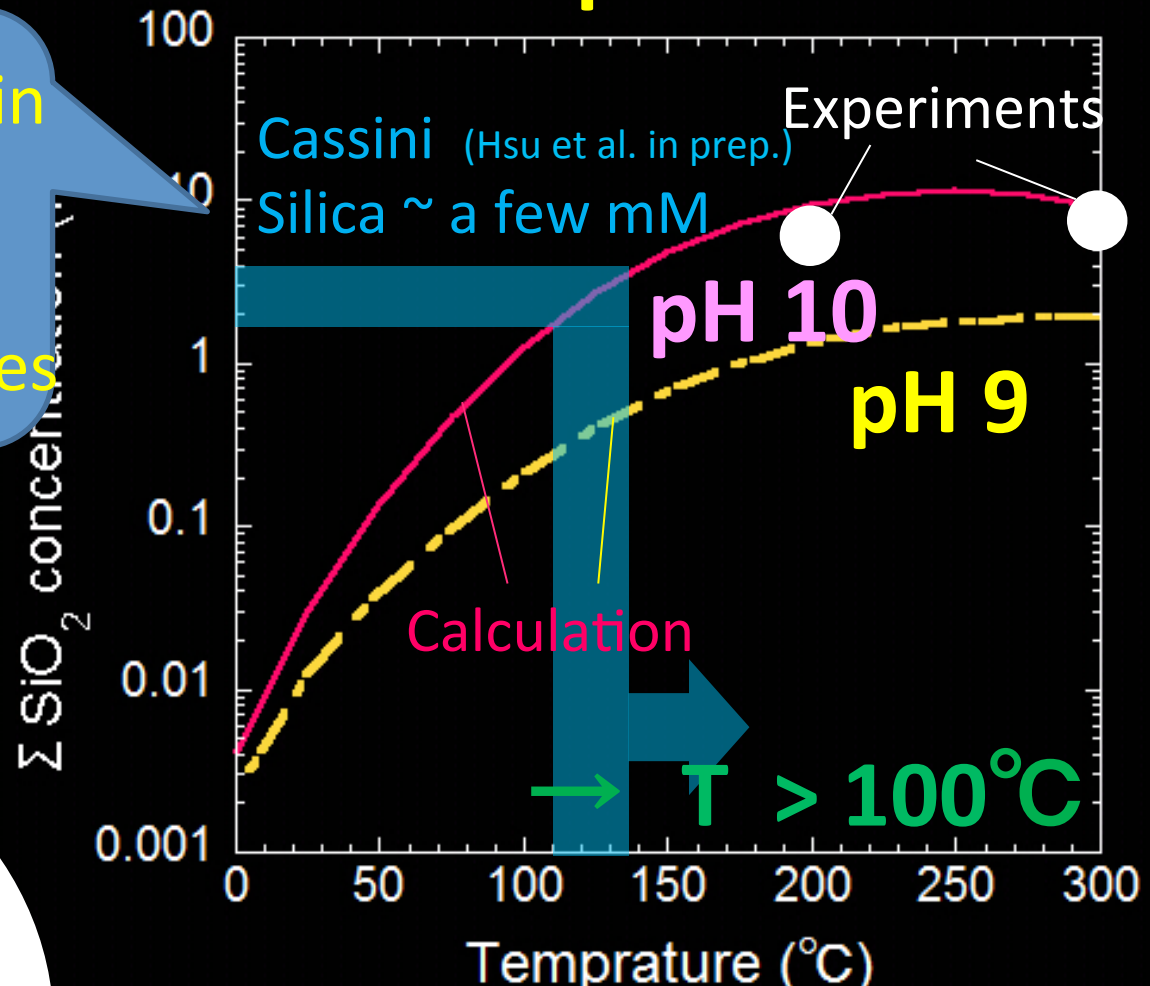
Fluid pH

Earth: mineral (e.g., brucite  $\Leftrightarrow$   $\text{Mg}^{2+} + 2\text{OH}^-$ )

Enceladus: volatiles (e.g.,  $\text{NH}_3 \Leftrightarrow \text{NH}_4^+ + \text{OH}^-$ )

→ pH 9–10

Silica concentration in fluids for pH 9 & 10 with alteration minerals of chondrites



Warm  
(~100-200 $^\circ\text{C}$ )

# More important result

planetary science + geochemistry + biology  
= oceanic planetology?



Ganymede



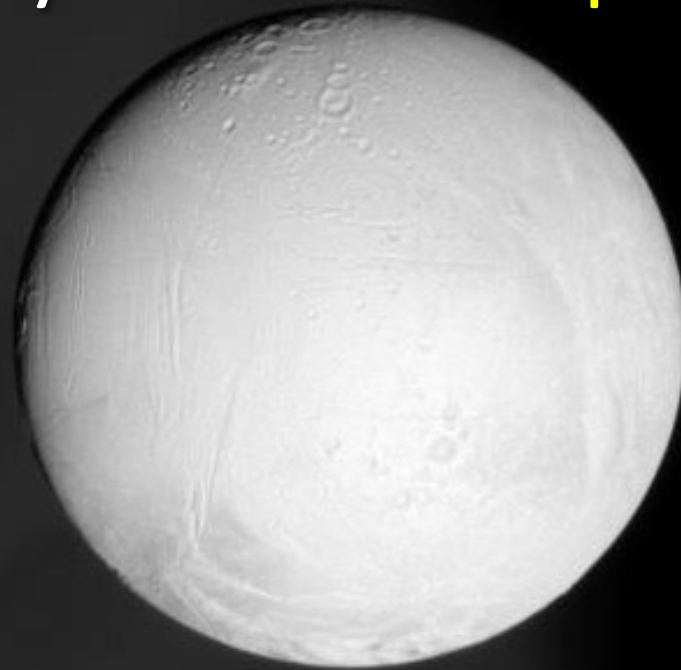
Europa



JUICE mission: detailed investigations of  
surface materials for Ganymede & Europa

# Conclusions

Icy satellites are a key for characterizing our solar system and geochemical processes and provide an opportunity for interdisciplinary researches.





# JUICE mission & Comparative ocean planetology



Europa

Low H<sub>2</sub>?

High temperature?

Ganymede

Callisto

Enceladus

Reducing?

High H<sub>2</sub>?

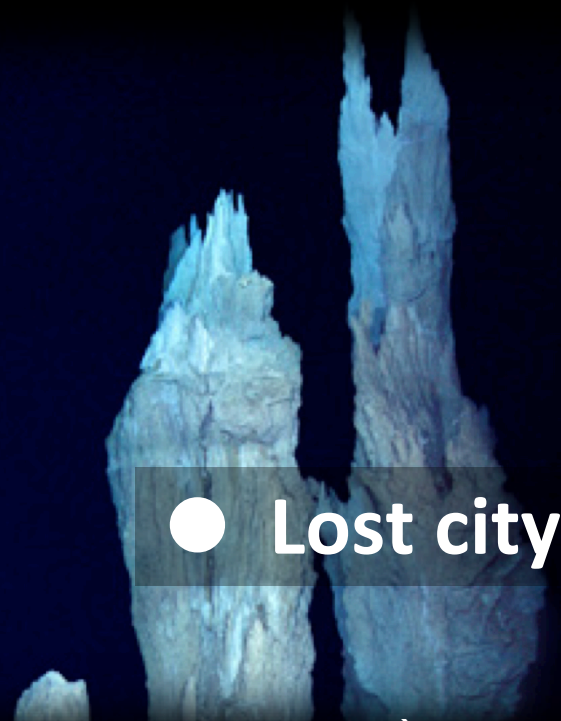
Earth



● Black smoker



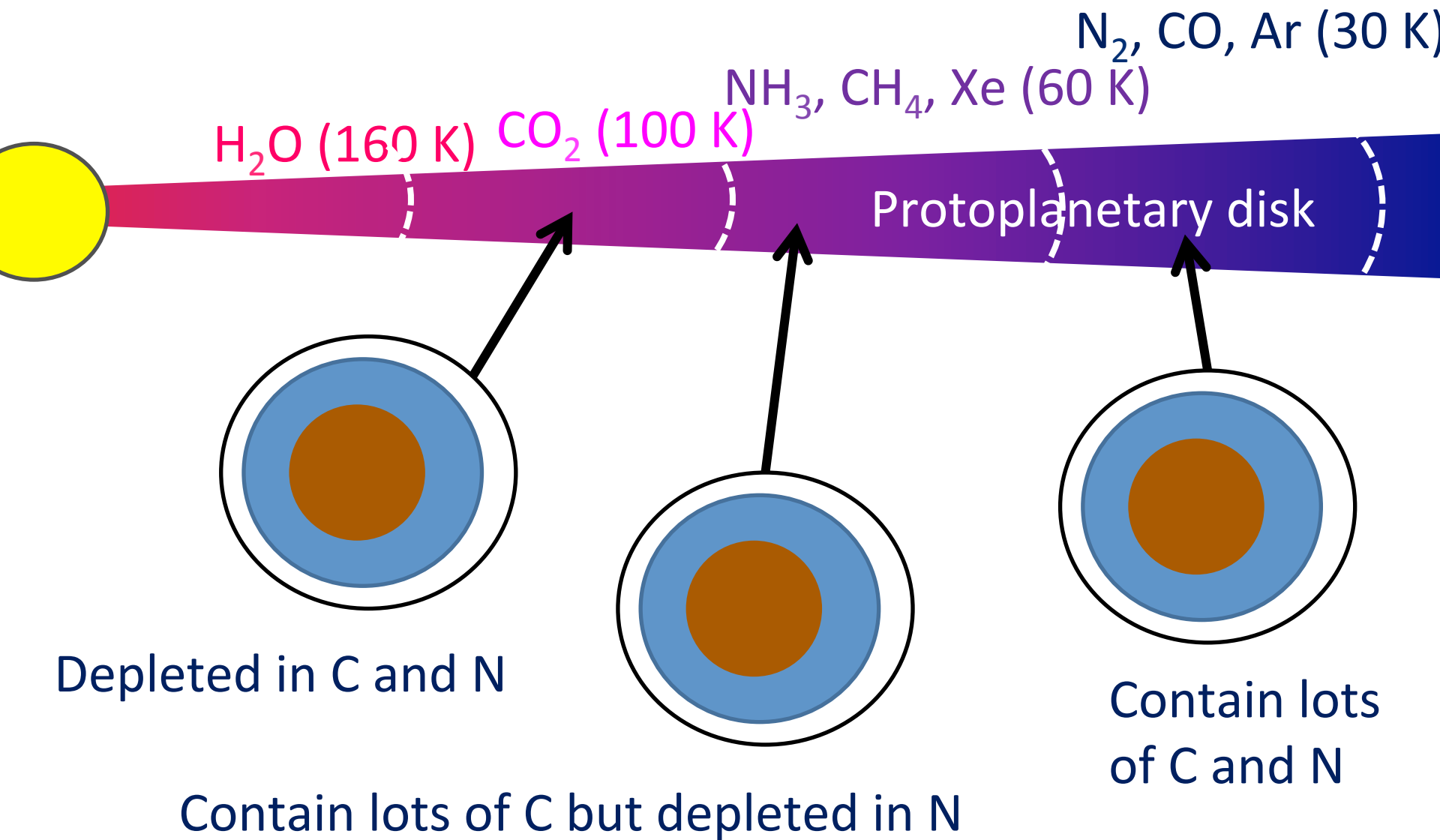
● Lost city



(Kelley et al., 2001; 2005; Proskurowski et al., 2008)



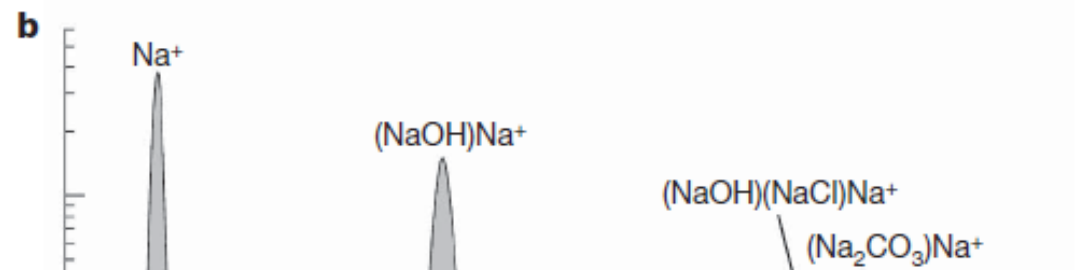
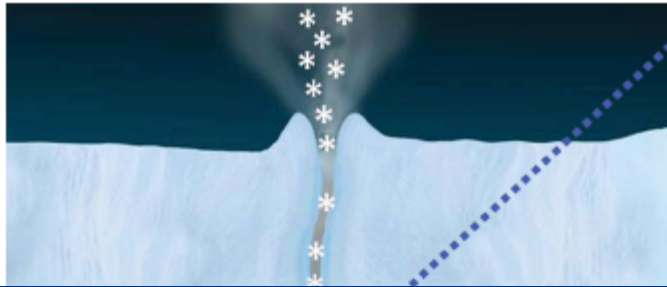
# Availability of elements in icy satellites



# Habitability

- Constraints on oceanic environments beneath the surface from chemical compounds of dust particles

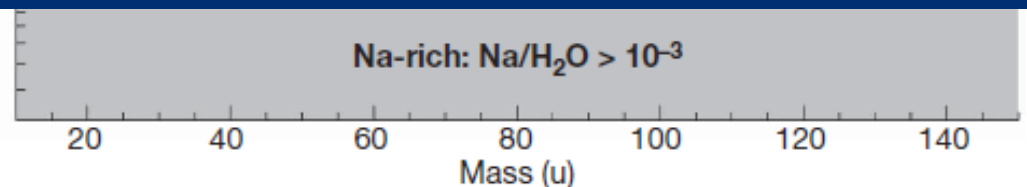
(Postberg et al., 2009; 2011)



**What are the compositions of Enceladus' ocean?**

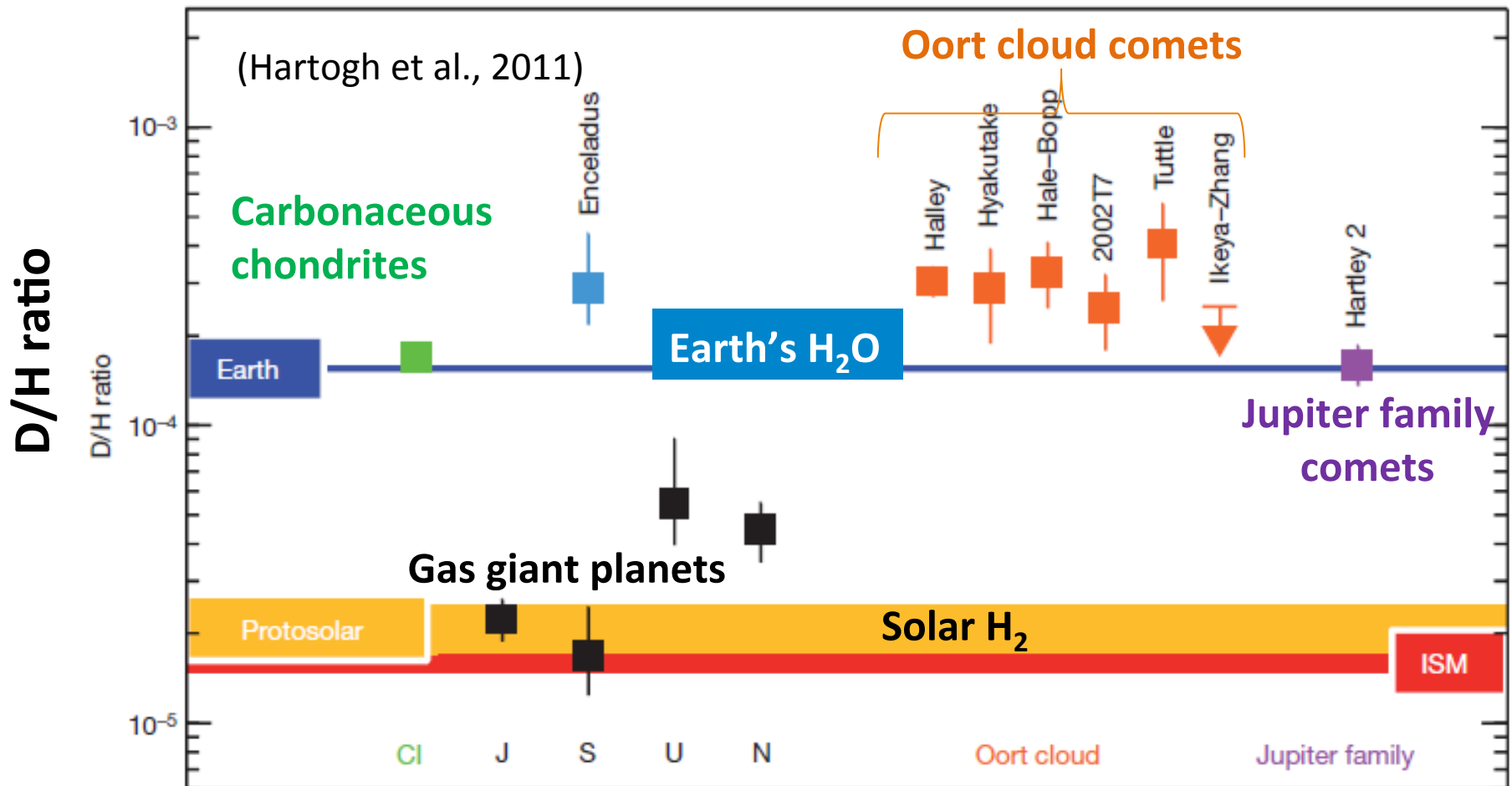
**What kinds of chemical reactions take place?**

- Dusty point of views from Cassini (Frank's & Sean's talk)
- From a biogeochemical point of view (Shibuya-san's talk)



# Origin & distribution of water

- Source(s) of H, C, O, and N → Isotopic compositions of primordial volatiles ( $\text{H}_2\text{O}$ ,  $\text{CO}$ ,  $\text{CO}_2$ ,  $\text{N}_2$ , &  $\text{NH}_3$ )
- Not a simple two component model



# Isotopes of volatiles

- Evolutional model of protoplanetary disk including isotopic exchanges (e.g., Drouart et al., 1999; Mousis et al., 2000; Yurimoto & Kuramoto, 2004)

Presolar

**D/H of H<sub>2</sub>O at Jupiter forming region**  
**⇒ Disk temperature & gas dissipation**

D/H evolution model in a disk

S